

INDIAN SCHOOL DARSAIT DEPARTMENT OF SCIENCE



S	Subject : Chemistry Topic : Metals and Non metals Date of Worksheet : 27.08.2019	
F	Resource Person: Mrs Sandhya Jitheesh Date :	
٢	Name of the Student : Class & Division : VIII Roll Number :	
1	Most metals react with dilute acids to produce :- a. hydrogen gas b. oxygen gas c. chlorine gas d. nitrogen gas	1
2	Fill in the blanks :-a. Non-metals are generally, so they break into pieces on hammering.b. Sodium and are soft enough to be cut with a knife.	1 Mark each
	c is the only non- metal which is a good conductor of electricity.d. Metal oxides are in nature because their solution in water turns red litmus blue.	
3	What would happen if sodium and potassium are kept in open air?	1
4	A reaction in which more reactive metal replaces a less reactive metal from its salt solution is called :- a. addition reaction b. double displacement reaction	1
5	 c. exothermic reaction Fill in the blanks :- a. Mercury is the only metal found in state at room temperature. b. Metals can be drawn into wires and this property is called c. The property of metals by which they can be hammered into very thin sheets without breaking is called d. Metals are conductors of heat and electricity. 	1 Mark each
6	Pick the odd one out of the following: Au/Ag/Pt/K/C	1
7 8	Cooking utensils are made of metals but their handles are made of wood or plastic, why? Why is it advised not to store pickles and curd in metallic utensils?	1 1
9	a. Name the products formed when a piece of Sodium metal is added to water.b. When the solution so formed, is tested with litmus, what kind of colour change is expected? Reason out your answer.	2
10	Can you guess?	1
	a. Non-metal essential for our life.b. Non-metal used in fertilizers for growth of plants.c. Non-metal which is used in purification of water.d. Non-metal used as an antiseptic over wounds.e. Non-metal used in fire crackers.	Mark each

11 Find out in which case reactions will occur and if so, the complete the equations.

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1 Mark

- b. CuSO₄ + Fe _____ c. $Cu + ZnSO_4 \longrightarrow$ d. $Cu + FeSO_4$ _____
 - e. $ZnSO_4 + Fe$ \longrightarrow
- 12 How do you test the presence of hydrogen gas?
- Give reasons:-13
 - a. When an iron nail is placed in copper sulphate Solution, the colour of the solution changes. each
 - b. Metals like sodium & potassium are stored in Kerosene oil while non-metals like phosphorous is stored in water.
 - c. Copper is used for making electrical wires.
 - d. Copper loses its lustre and develops a green coating over it when exposed to atmosphere.